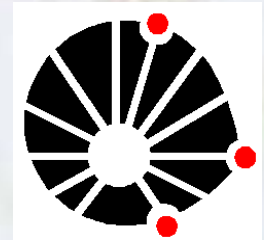


Microstructure and Mechanical Properties of Directionally Solidified Ti-Fe Eutectic Alloy

Rodrigo Contieri, Eder Lopes and
Rubens Caram
University of Campinas, Brazil

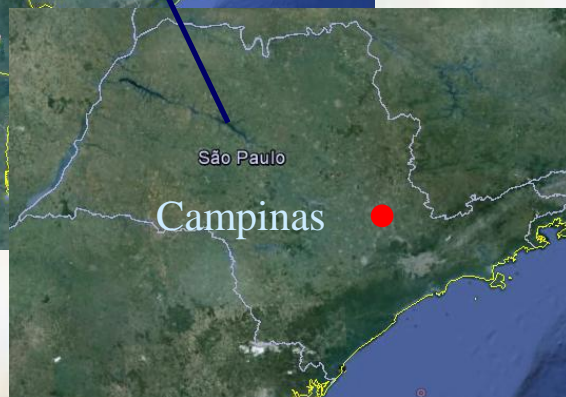


4th international
DSEC

Washington, DC
October, 2012



Campinas, SP, Brazil





University of Campinas

- **Founded in 1966**
- **Strong tradition in education and in scientific research (15% of the Brazilian Scientific Production)**
- **17,000 undergraduate and 16,000 graduate students**





Outline

- **Eutectic Alloys**
- **Eutectic Growth in Metallic Systems**
- **Eutectic Arrangements**
 - **Previous Studies:**
 - **Ni-Si; Al-Nb; Ni-Al-Mo; Al-Nb-Ni**
- **Ti-Fe Eutectic Alloy**
- **Results – D.S. of Ti-Fe Eutectic Alloy**
 - **DS based on Arc Melting Equipment**
 - **Microstructure**
 - **Mechanical Behavior**
 - **Conclusions**



Eutectic Alloys

- Eutectic alloys allow the developing of in situ composite for structural applications
- This material consists of phases embedded in a matrix that do not dissolve in each other and are physically separated by a sharp interface between them
- This composite material provides the opportunity of merging the properties of distinct constituents into one material.



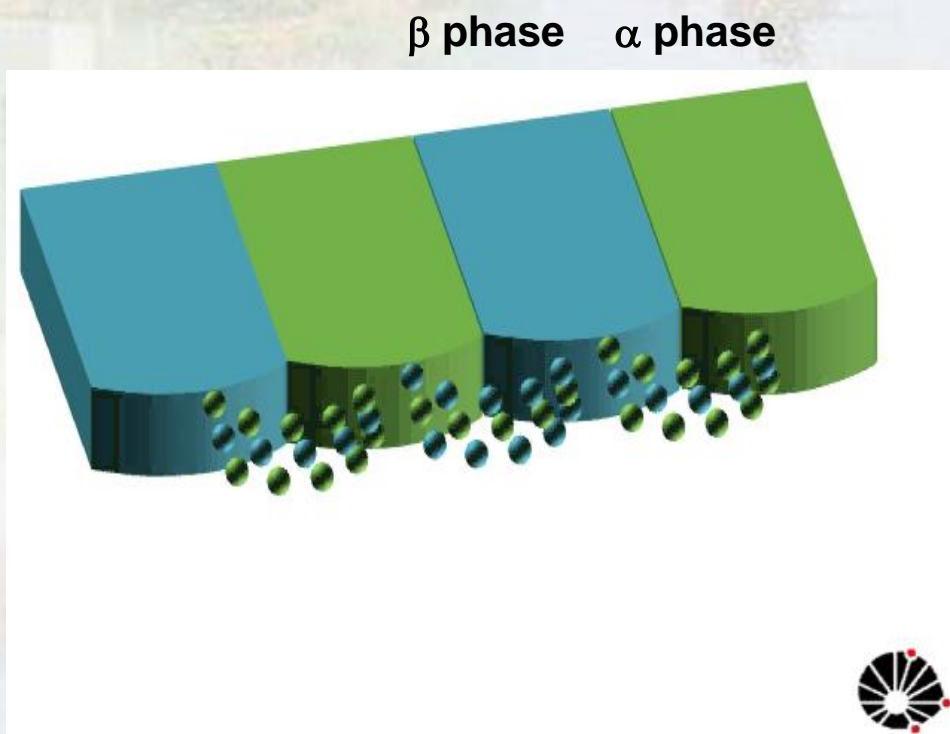
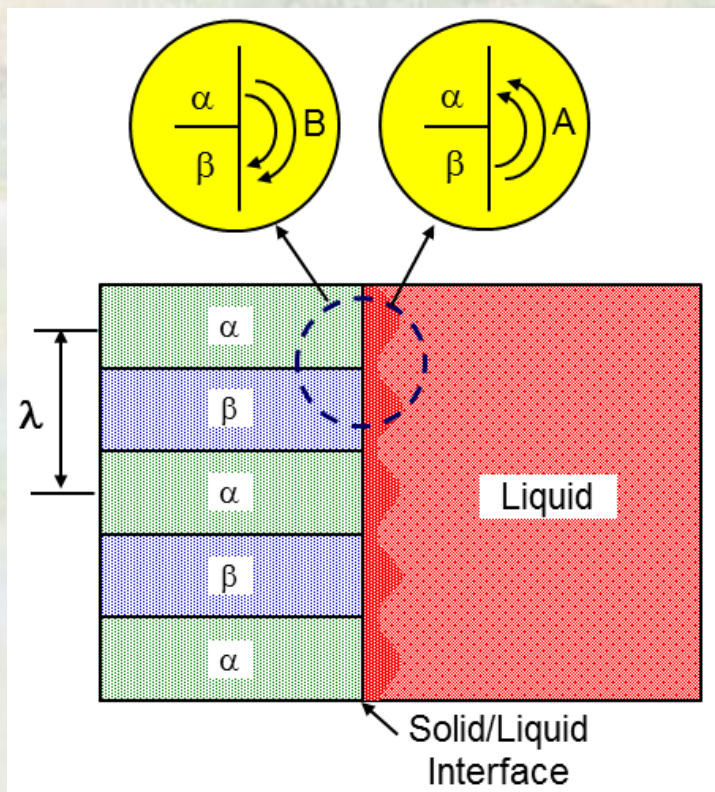
Eutectic Growth

- **Growth of eutectic alloys is an effective method in obtaining in situ composite materials**
- **In situ composites generally have a high degree of thermal stability and improved mechanical properties**
- **D.S. eutectic alloys results in regular structure of two or more solid phases**
- **Eutectic solidification leads to cooperative growth**



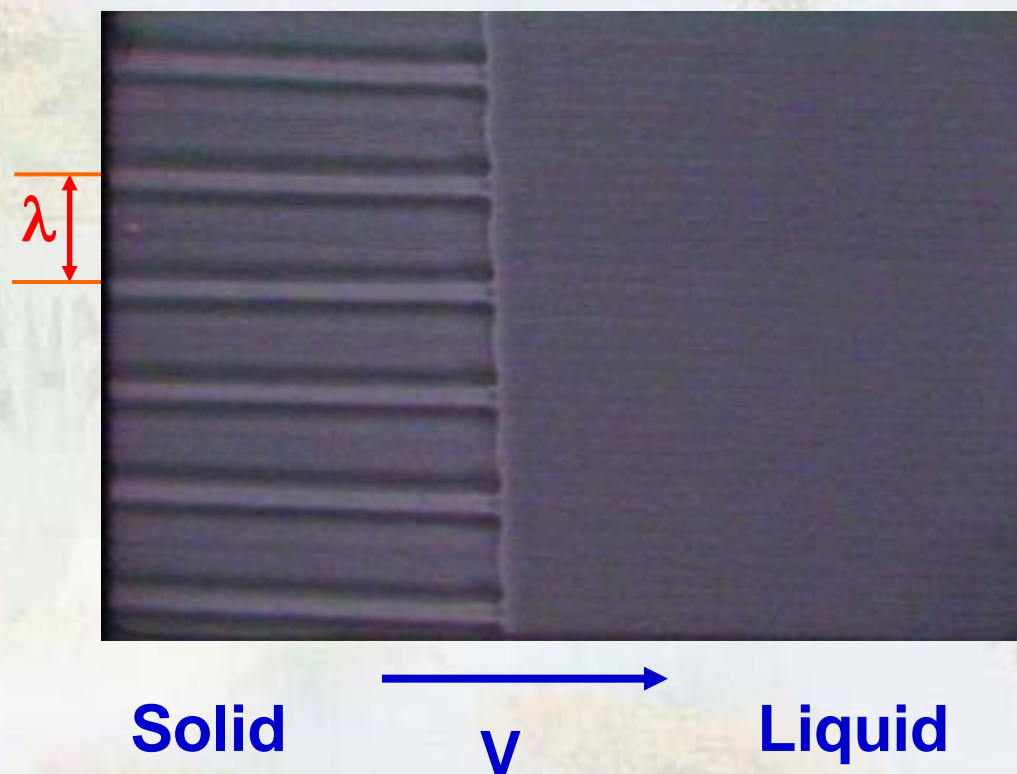
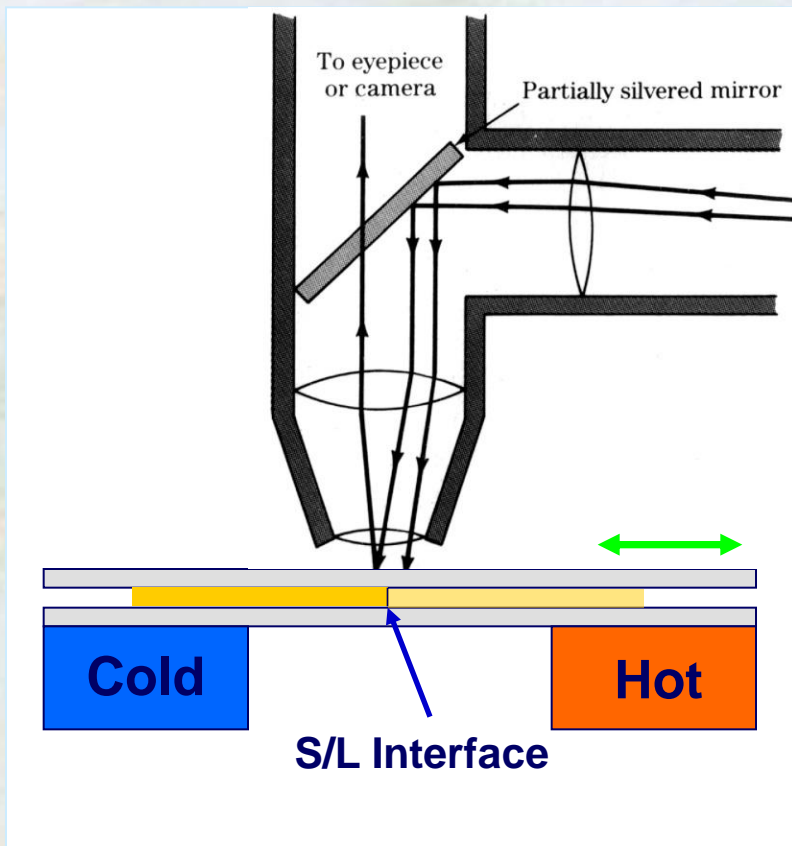
Cooperative Growth

While the α phase segregates B, the β phase rejects A
Such a phenomenon leads to a solute build up in the liquid in front of the α e β phases and hence, to lateral solute diffusion of A and B



Eutectic Growth

Growth of $\text{CBr}_4\text{-C}_2\text{Cl}_6$ eutectic organic alloy



J.D. Hunt and K.A. Jackson – Bell Laboratories – 60's



Previous Studies

Ni-Ni₃Si - Lamellar

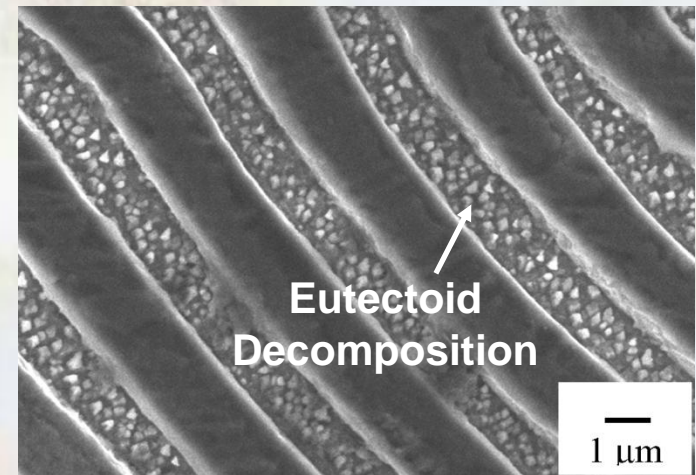
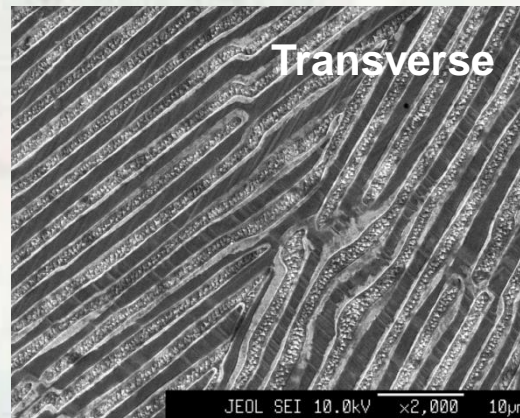
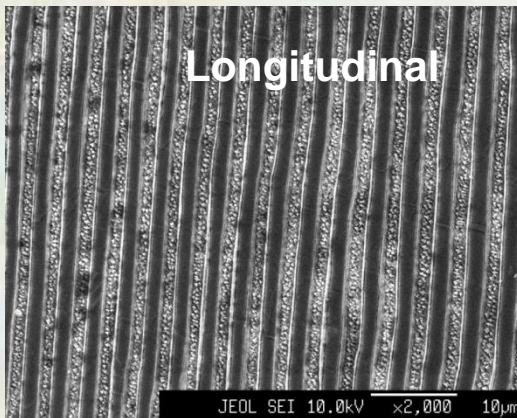
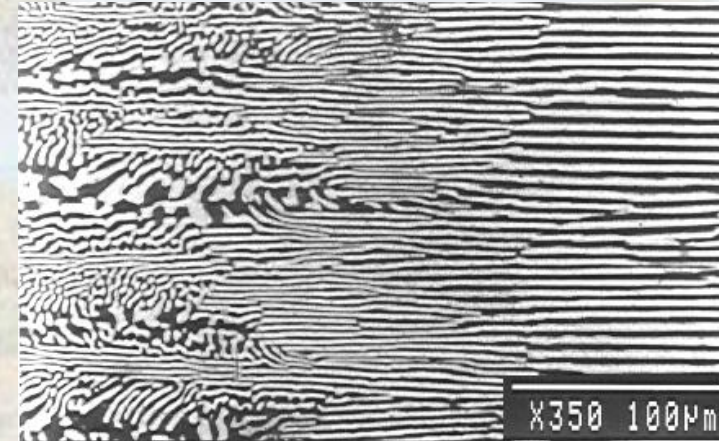
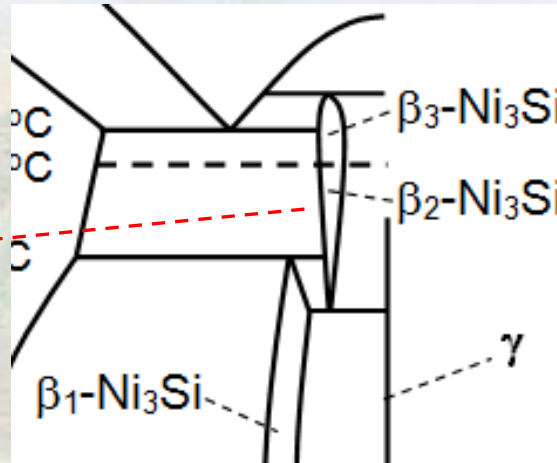
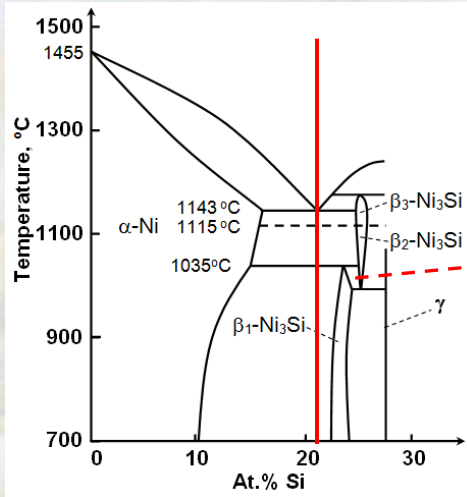
Al₃Nb-Nb₂Al - Lamellar

Ni-Al-Mo - Rod-like

Al₃Nb-Nb₂Al-AlNiNb - Ternary



Lamellar Eutectic Growth

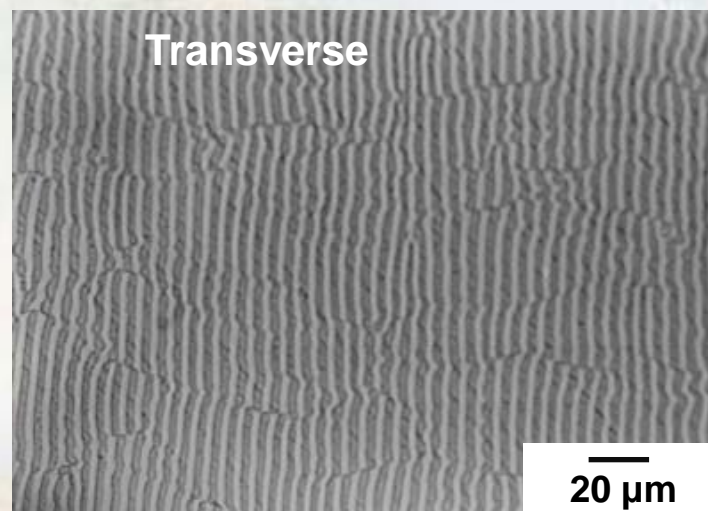
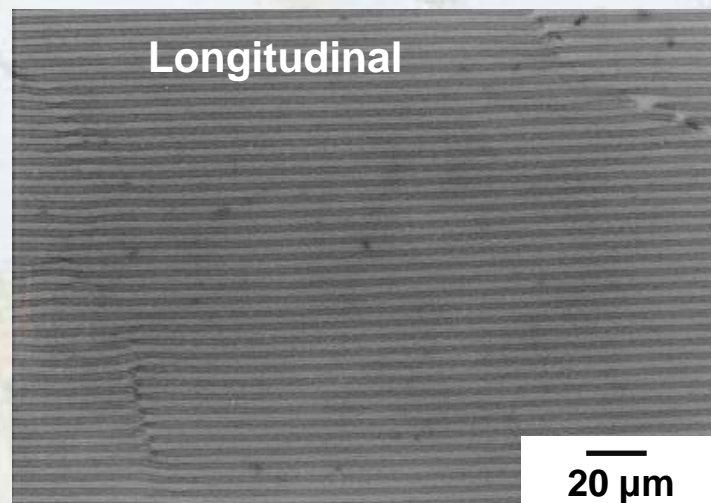
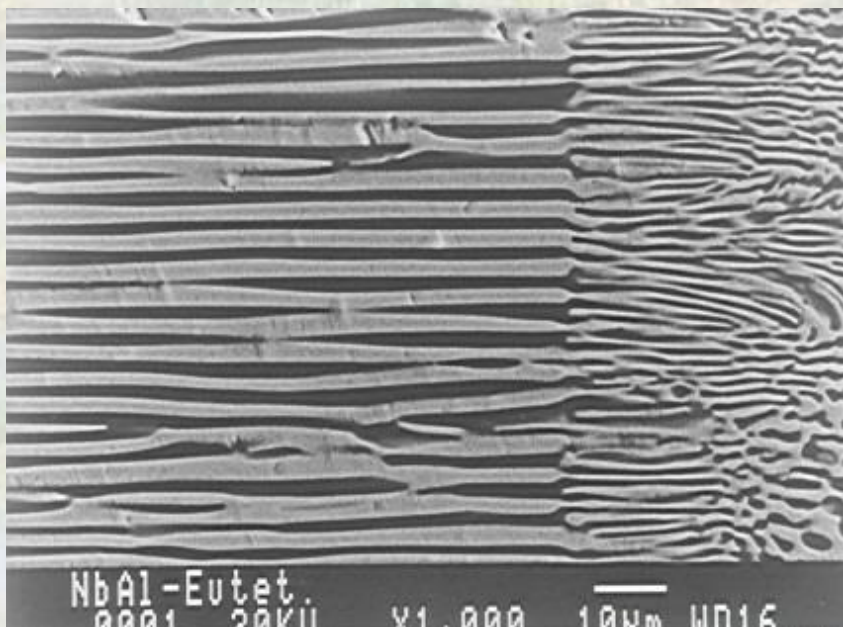




Lamellar Eutectic Growth



Eutectic Transformation
1595°C / Al-42.2Nb at%

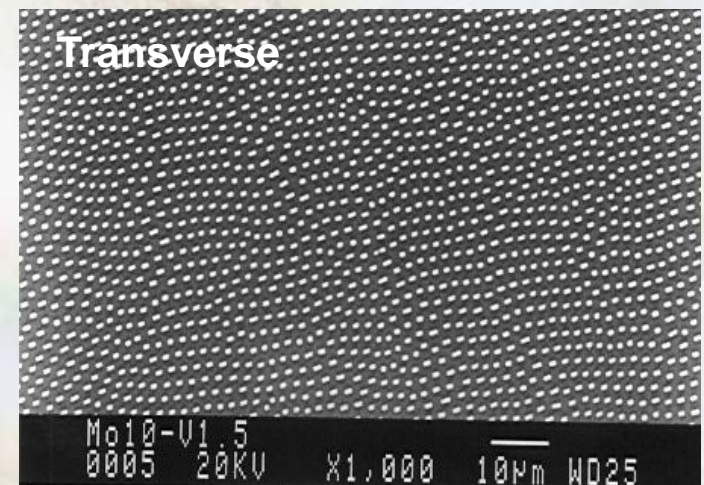
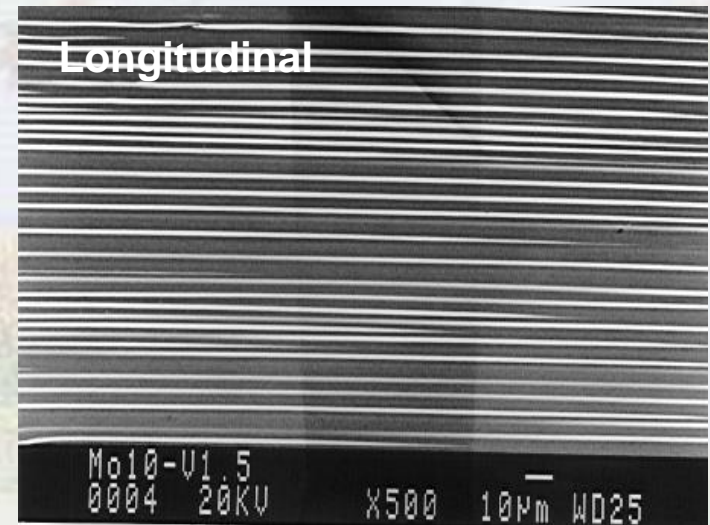
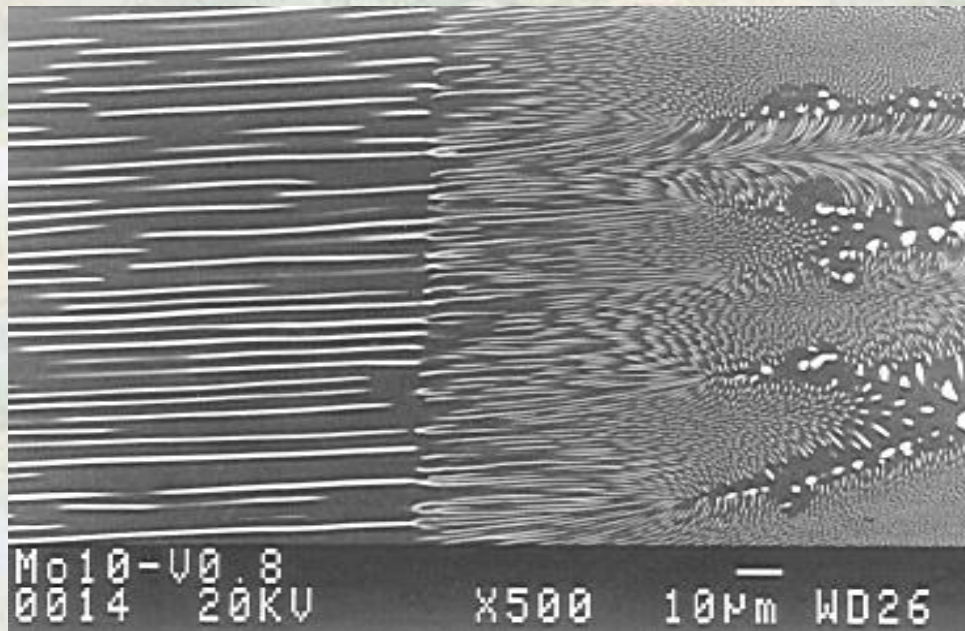




Rod-Like Eutectic Growth

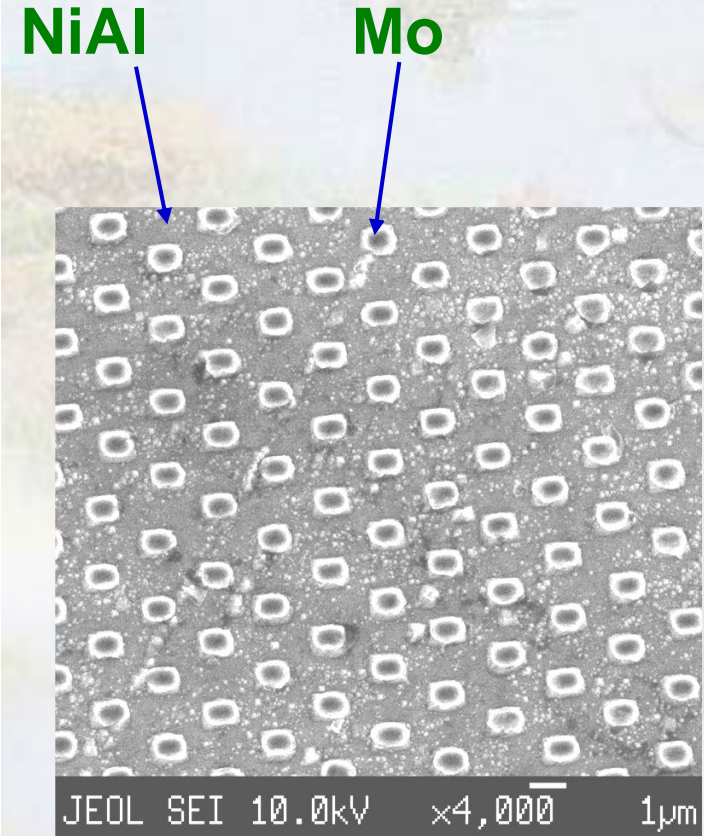
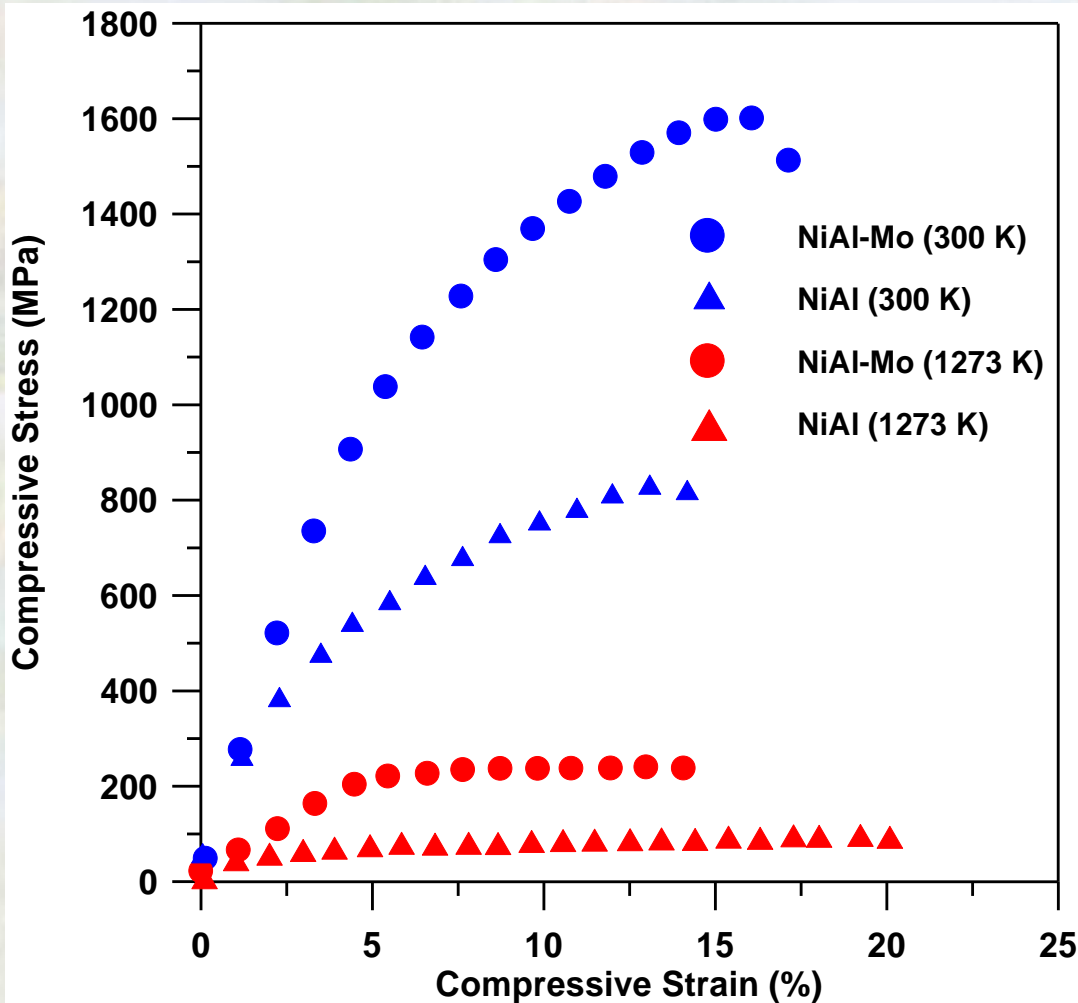


Eutectic Transformation
1600°C / NiAl-10Mo at%



Rod-Like Eutectic Growth

NiAl-Mo In Situ Composite

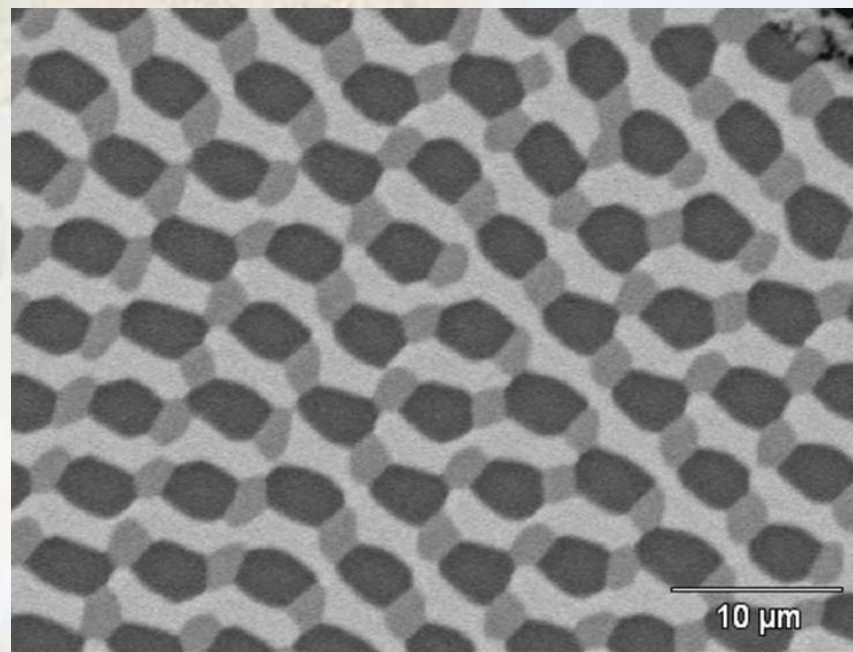
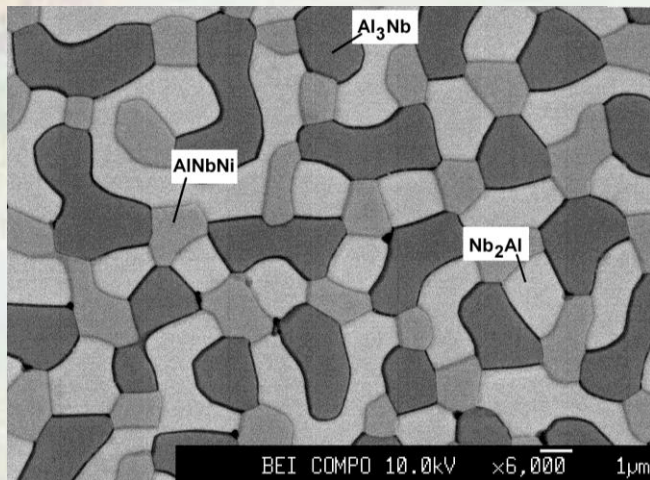
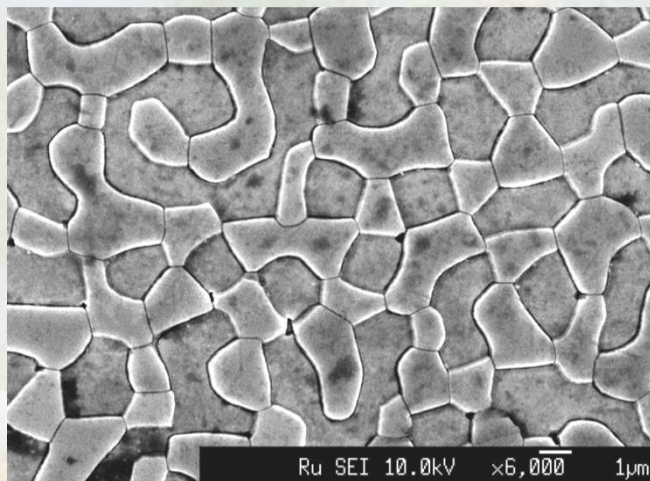




Ternary Eutectic Growth



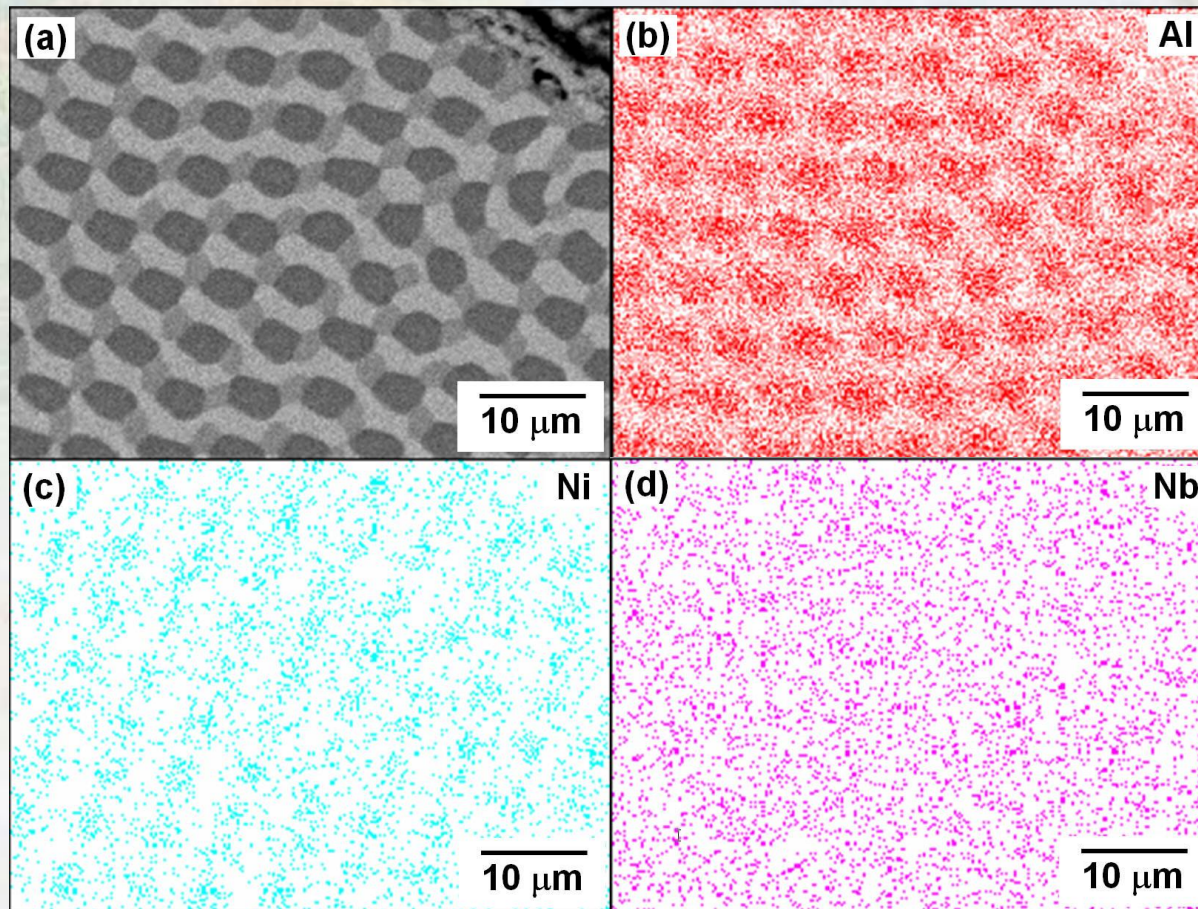
Eutectic Transformation 1520°C / Al-40.4Nb-2.4Ni at%



Scripta Materialia 48 (2003) 1495

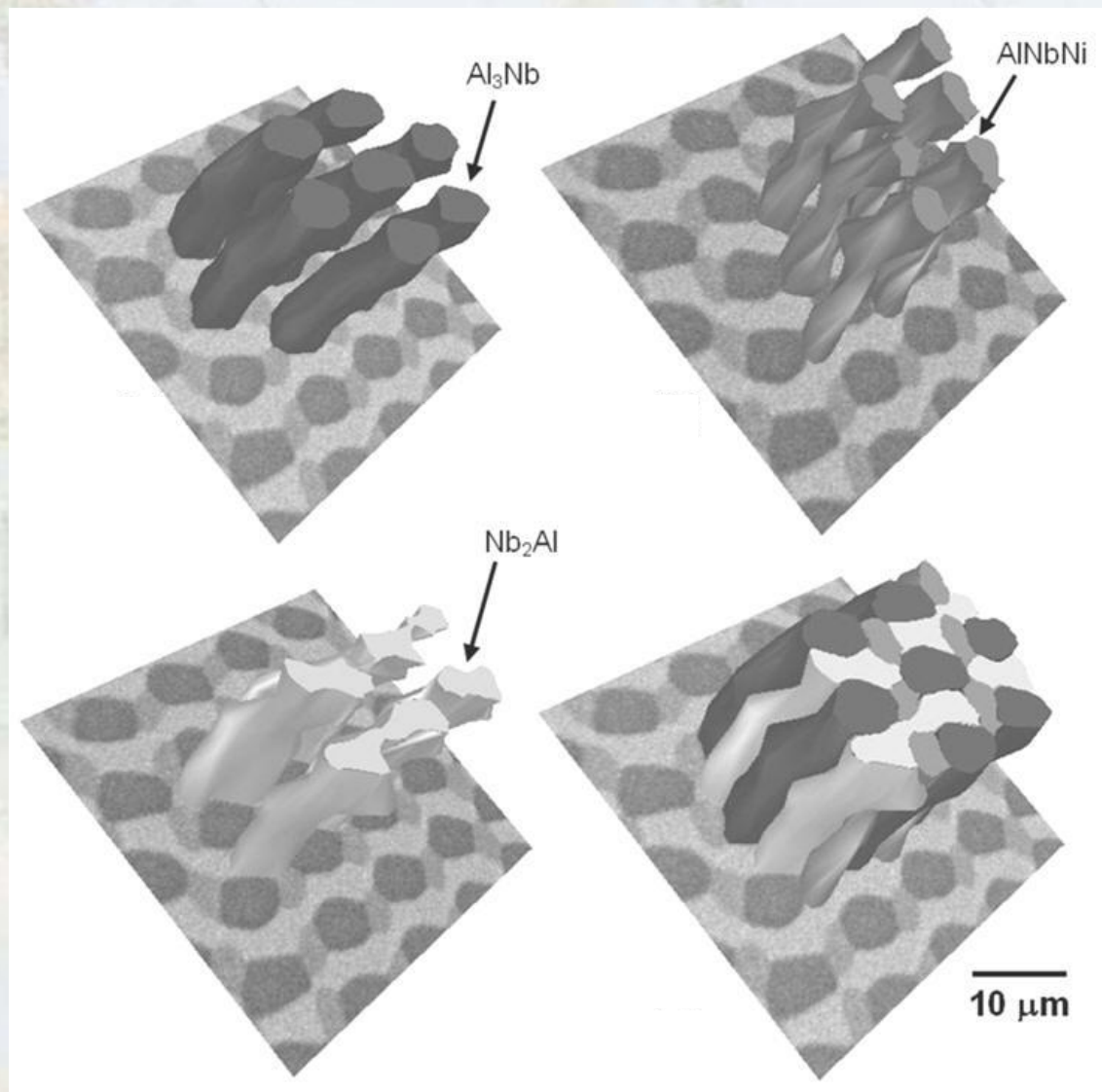
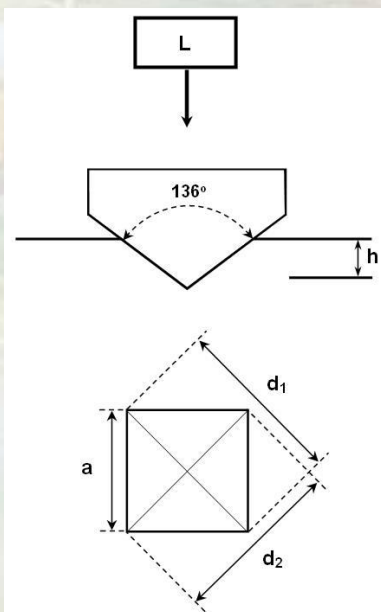
Ternary Eutectic Growth

Atom distribution by X-Ray Maps



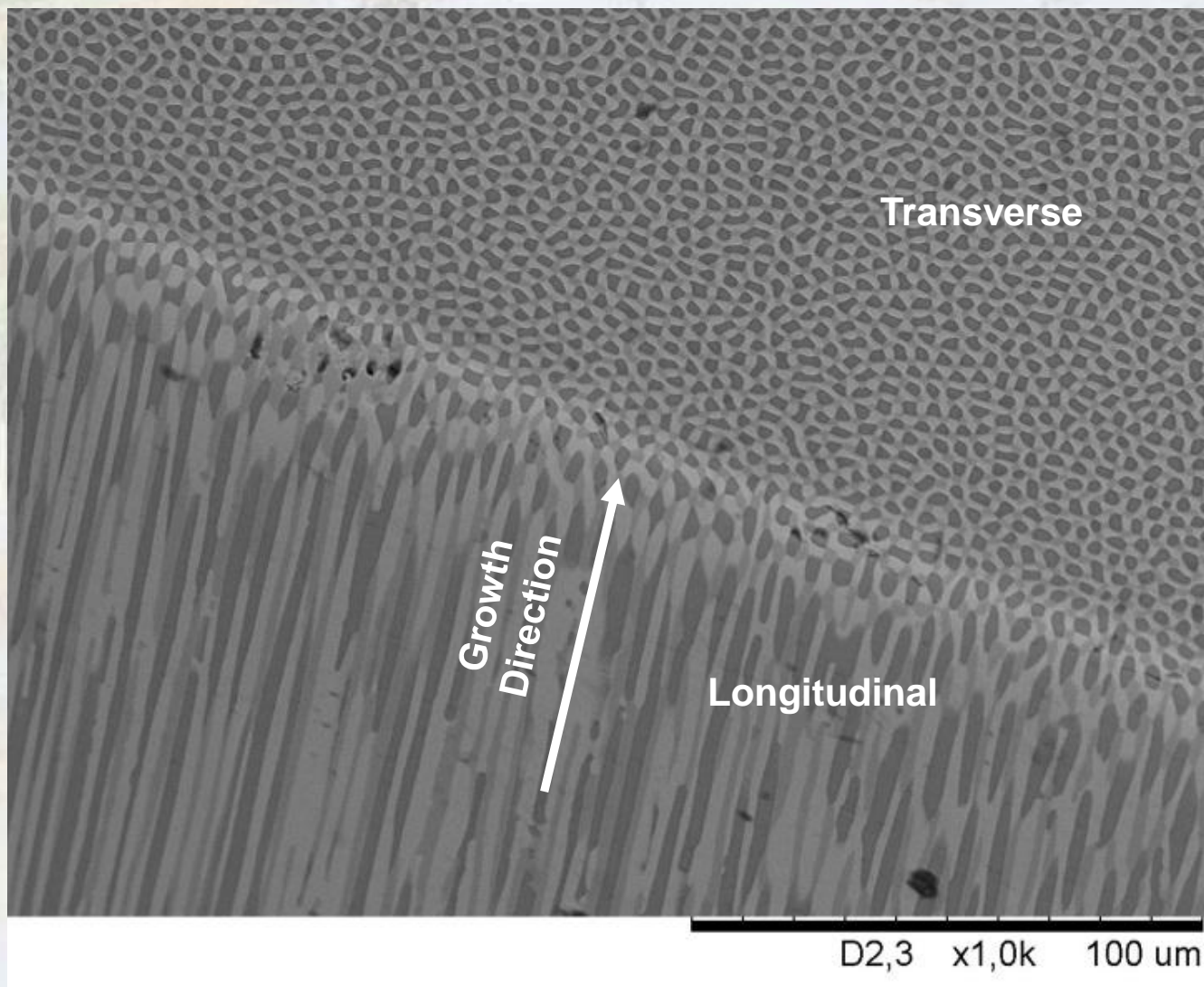
Ternary Eutectic Growth

3D reconstruction
of the ternary
eutectic
microstructure
using the serial
sectioning
technique





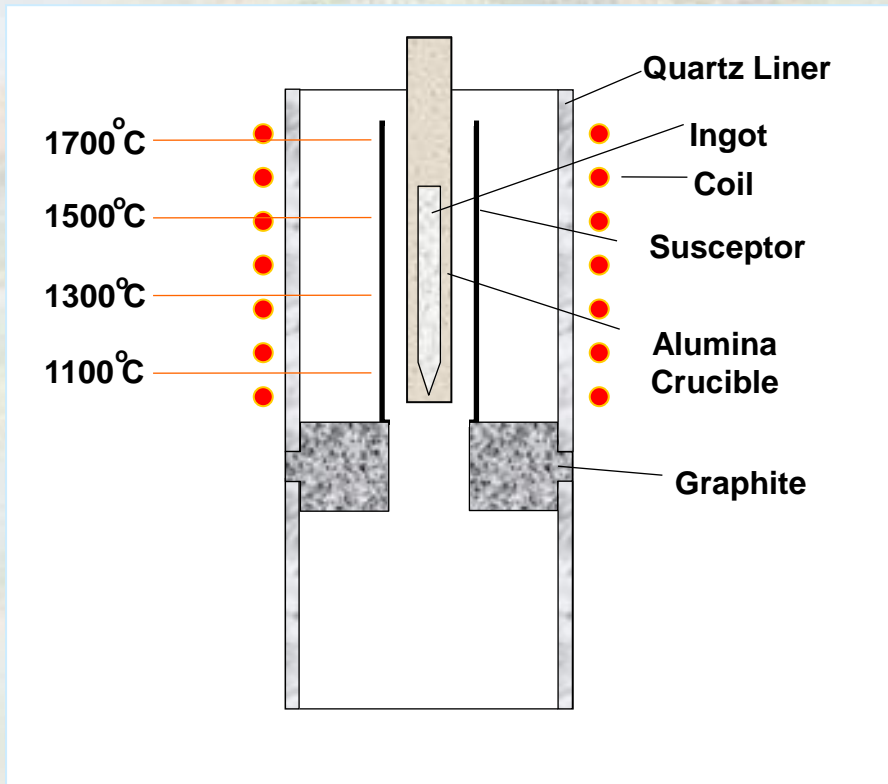
Ternary Eutectic Growth



Materials & Design 33 (2012) 563

Directional Solidification

- D.S. of eutectics was carried out by using a vertical Bridgman furnace, in Al_2O_3 crucibles (0.8 ID x 1.0 cm OD and 6.0 cm long)

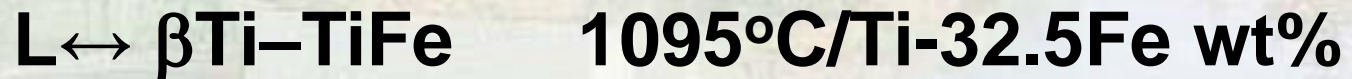


- **Ti alloys can not be processed in Al_2O_3 crucibles**



Ti-Fe System

- Mechanical performance of Ti can be considerably enhanced by combining it and Fe, causing an eutectic transformation:



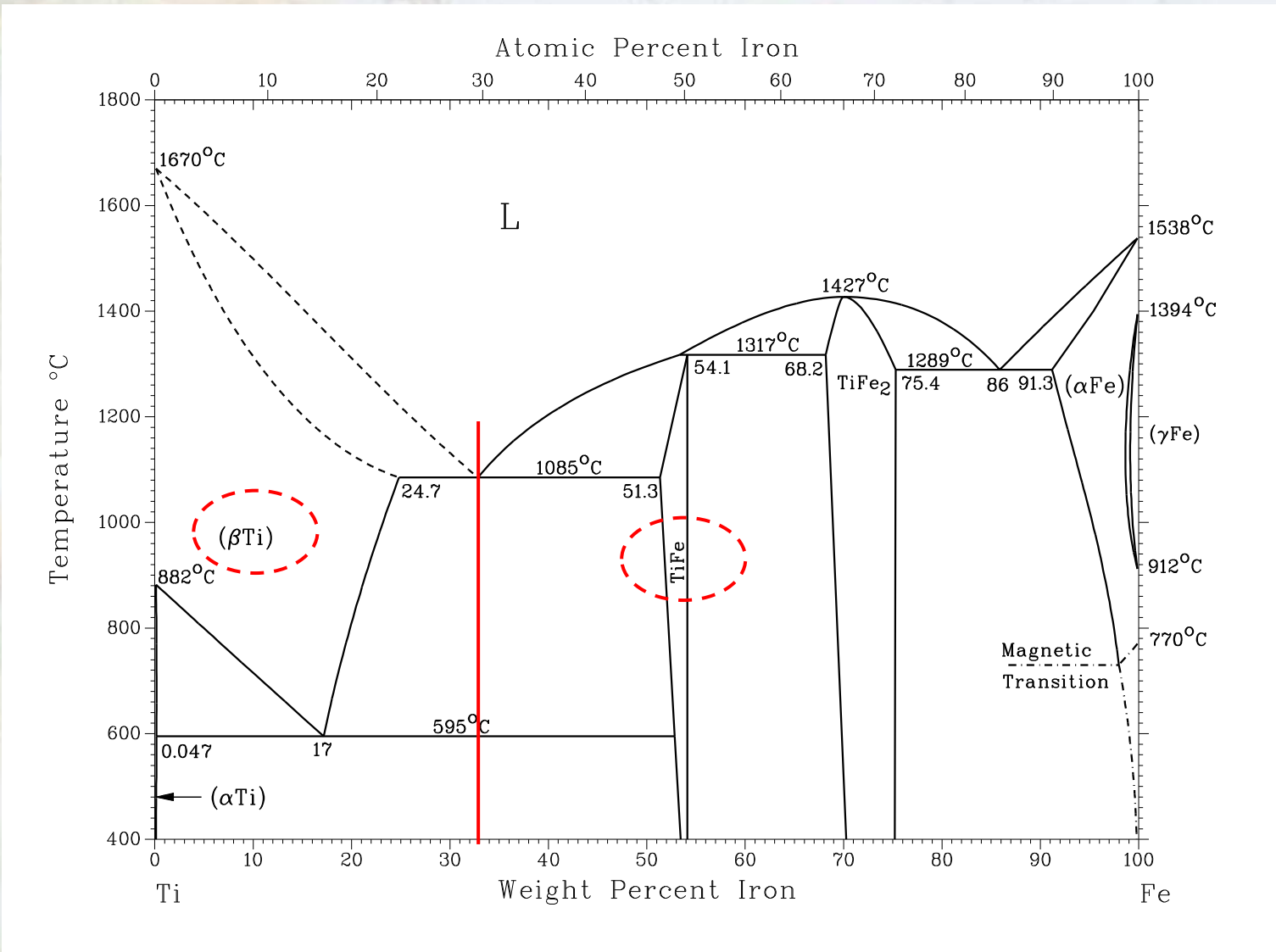
βTi : ductile BCC phase

TiFe: high strength phase

- Directional solidification was carried out in a setup that employs a water-cooled copper crucible combined with a voltaic electric arc moving through the sample.



Ti-Fe System

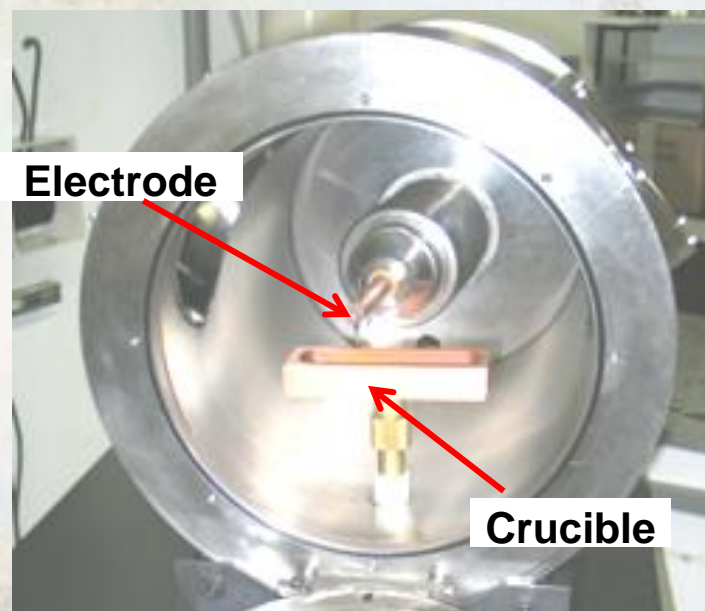


Sample Preparation

- Arc furnace with non-consumable W electrode and water cooled copper hearth under Ar atmosphere.



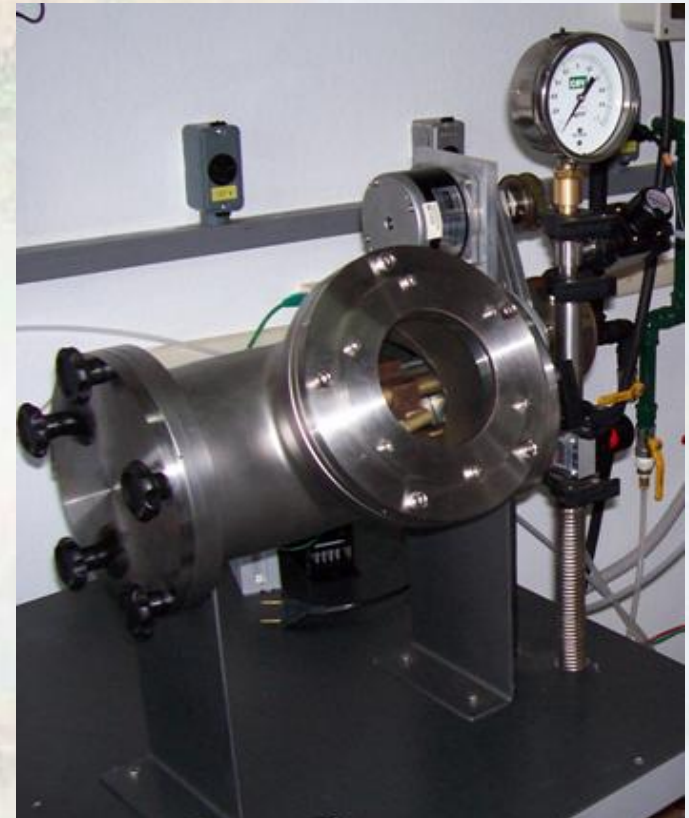
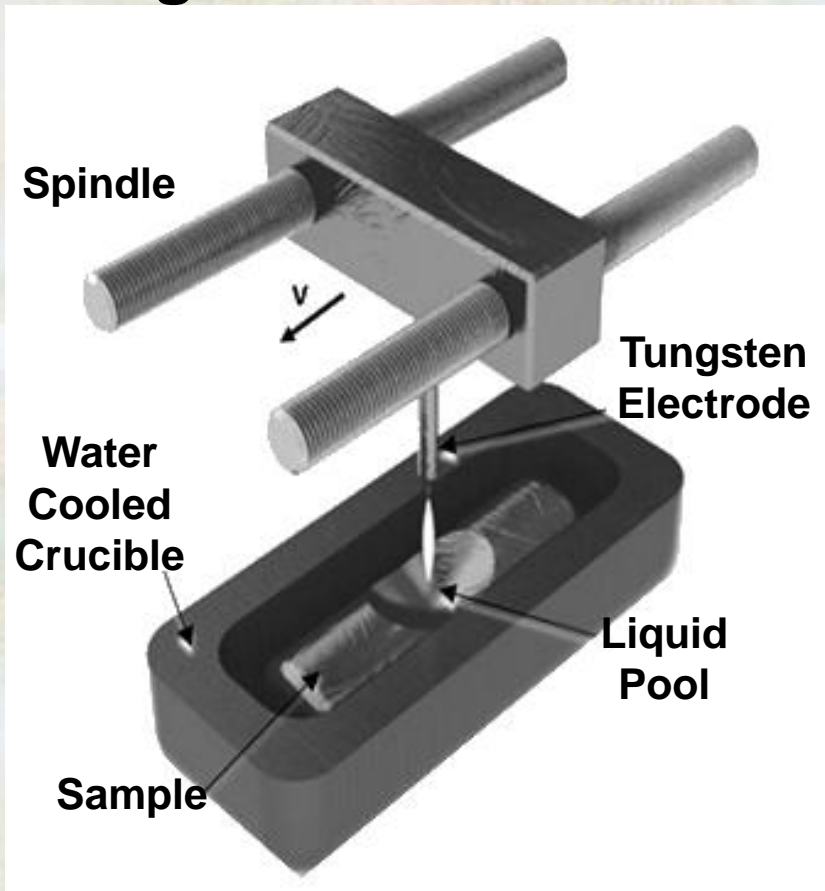
Nominal (at.%)
Ti-32.5 Fe





Directional Solidification

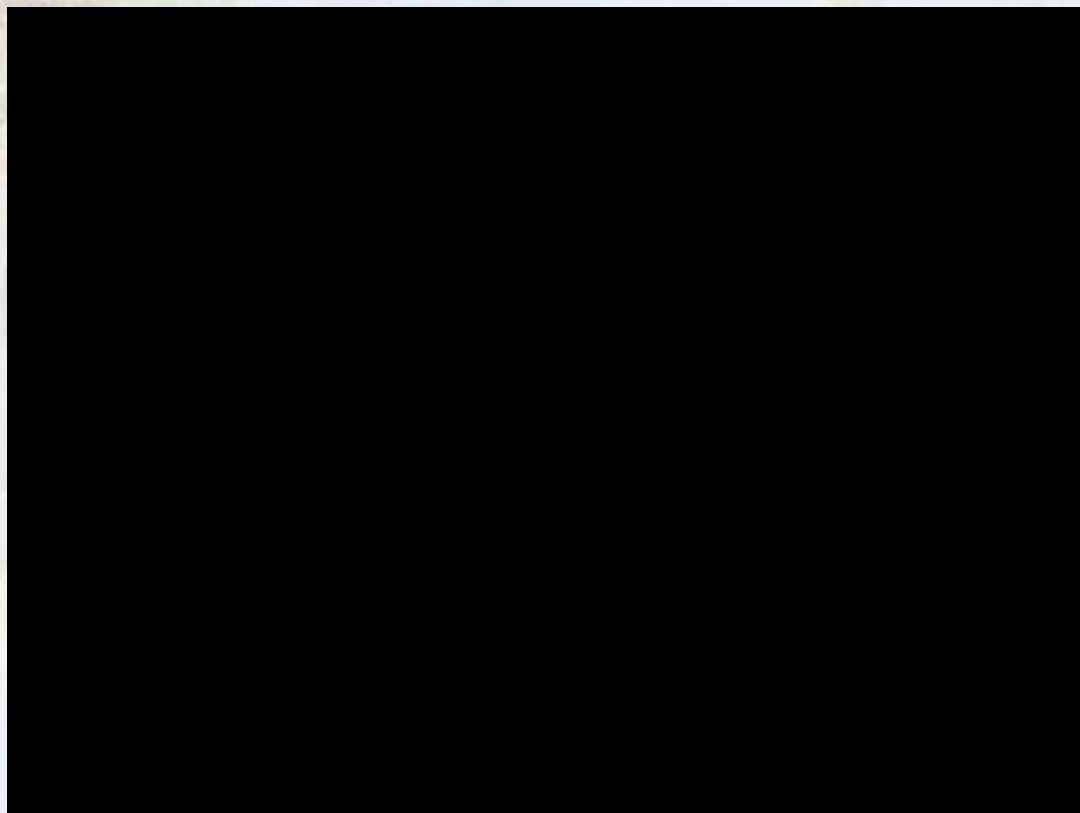
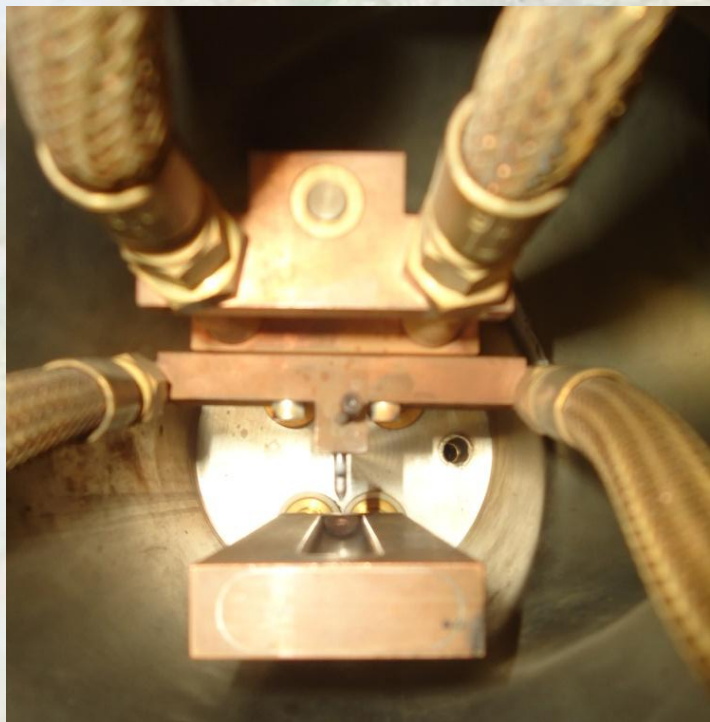
- Arc furnace with a nonconsumable W electrode that moves longitudinally along the ingot at different rates





Directional Solidification

- Three solidification rates chosen:
 $V=10, 30$ and 60 mm/h.





Sample Characterization

- **Chemical composition**
 - X-ray fluorescence spectrometry - Rigaku RIX 3100
 - Oxygen and nitrogen - LECO TC-400 analyzer
- **Phase transformations**
 - Differential thermal analysis - Netzsch STA 409
- **Microstructure characterization**
 - Scanning electron microscopy - Zeiss EVO 15
 - Transmission electron microscopy - JEOL JEM 2100
 - X-ray diffraction - PANalytical X'Pert
- **Mechanical characterization**
 - Vickers Hardness test – Buehler 2100
 - Nano-indentation – NHT – CSM Instruments
 - Compressive tests – EMIC DL2000



Chemical Composition

- Chemical composition:
 - X-ray fluorescence spectrometry - Rigaku RIX 3100
 - Oxygen and nitrogen - LECO TC-400 analyzer

Nominal (at.%)	Measured (at.%)
Ti-32.5 Fe	Ti-32.8 Fe

Ti	O (wt.%)	N (wt.%)
Balance	0.0855	0.014725

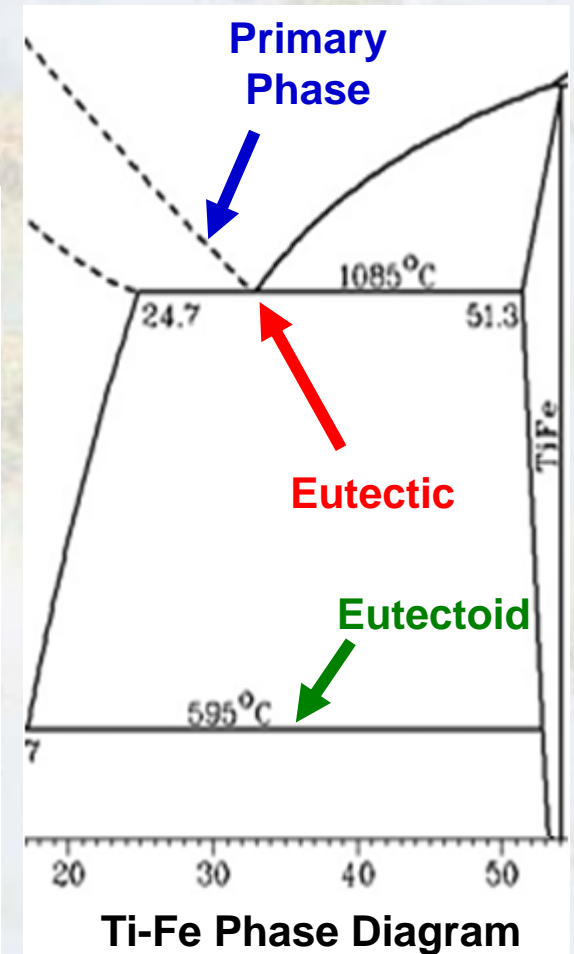
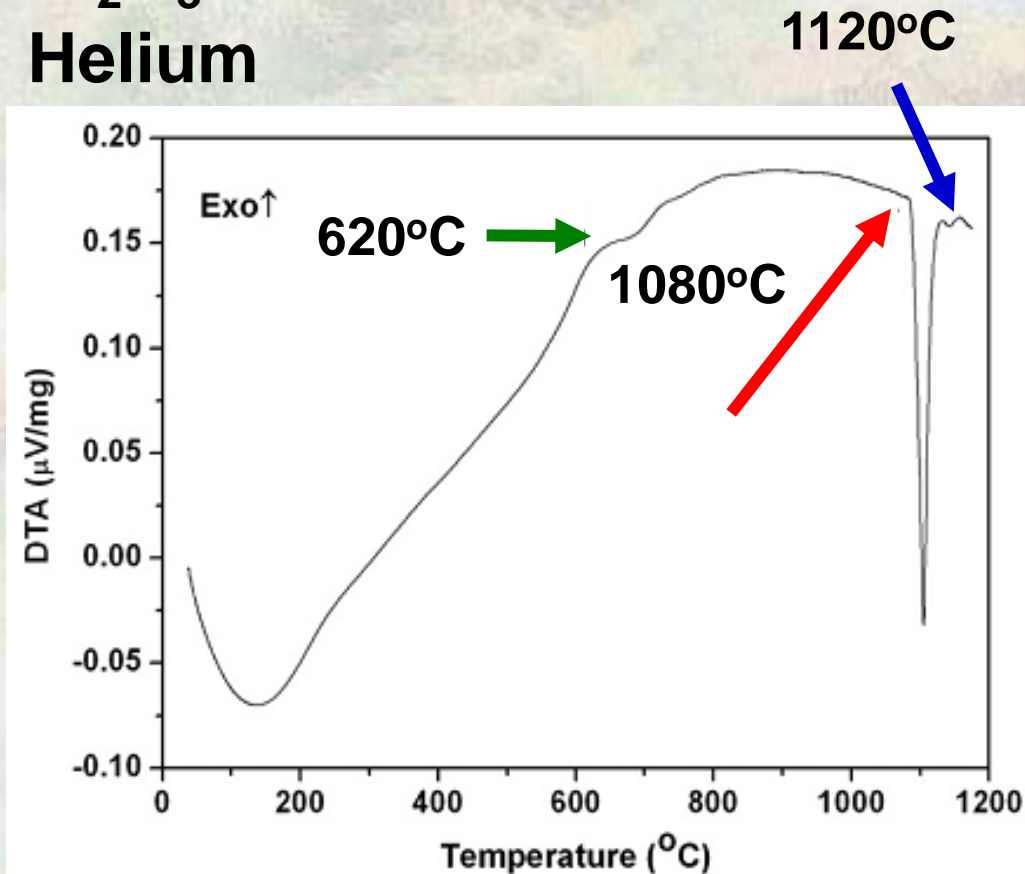
**very low interstitial
contamination**





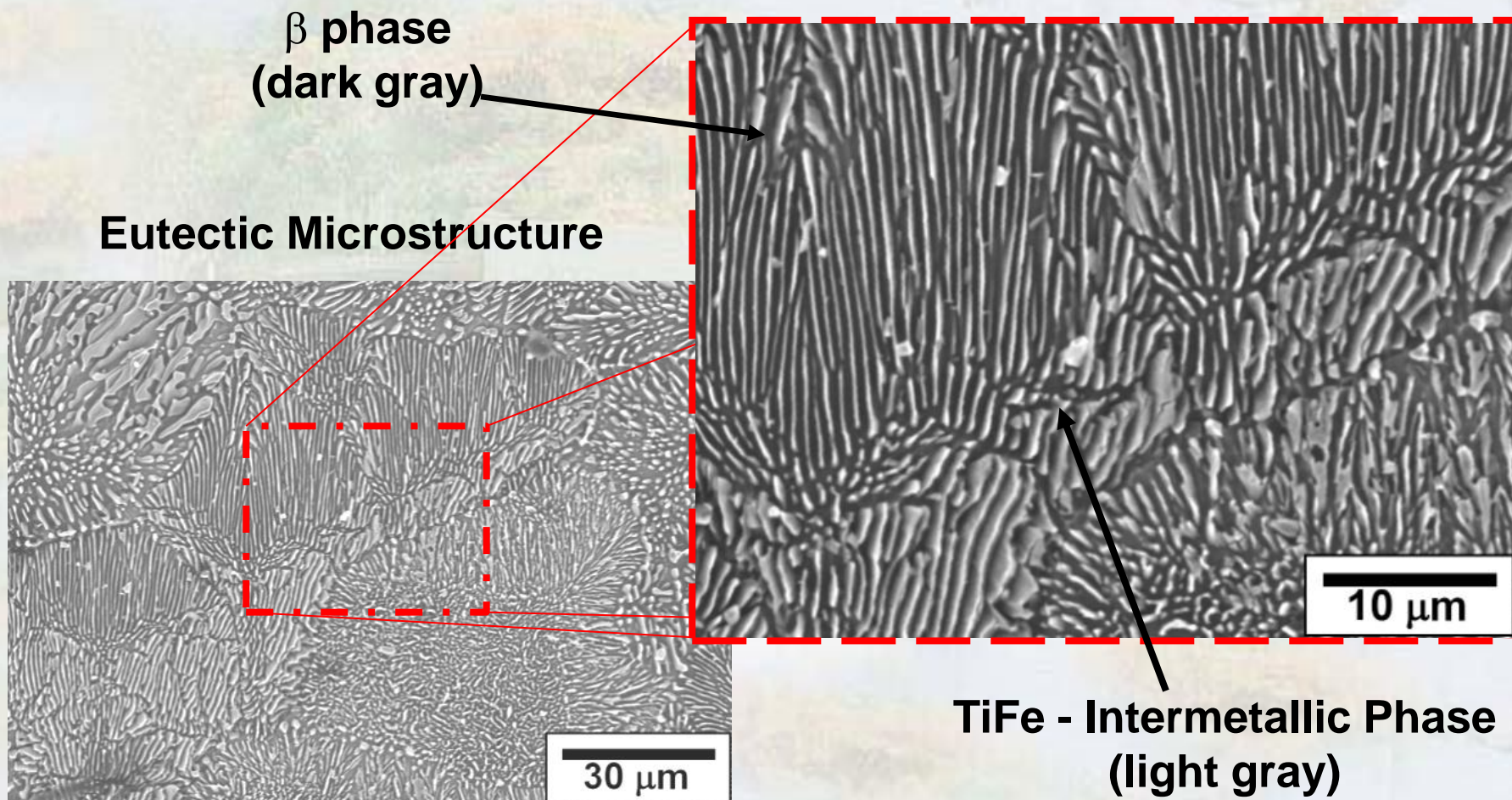
Phase Transformations

- DTA - Ti-32.5Fe eutectic alloy
- Heating rate of 10°C/min
- Al₂O₃ Crucible
- Helium



As-Cast Condition

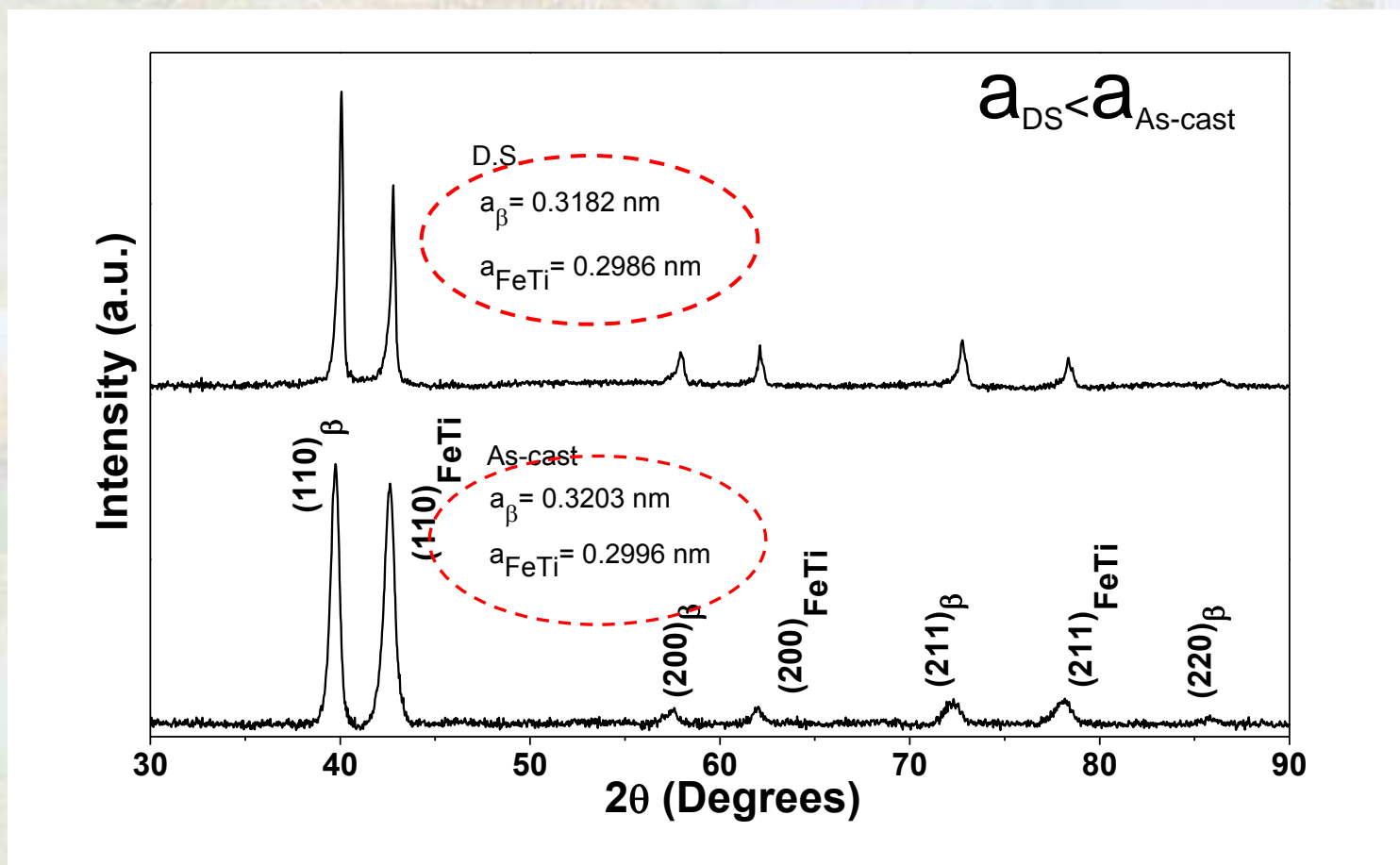
- SEM micrographs of the Ti-Fe eutectic alloy in the as-cast condition





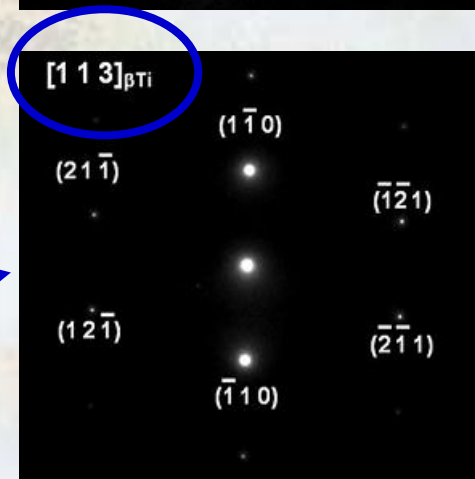
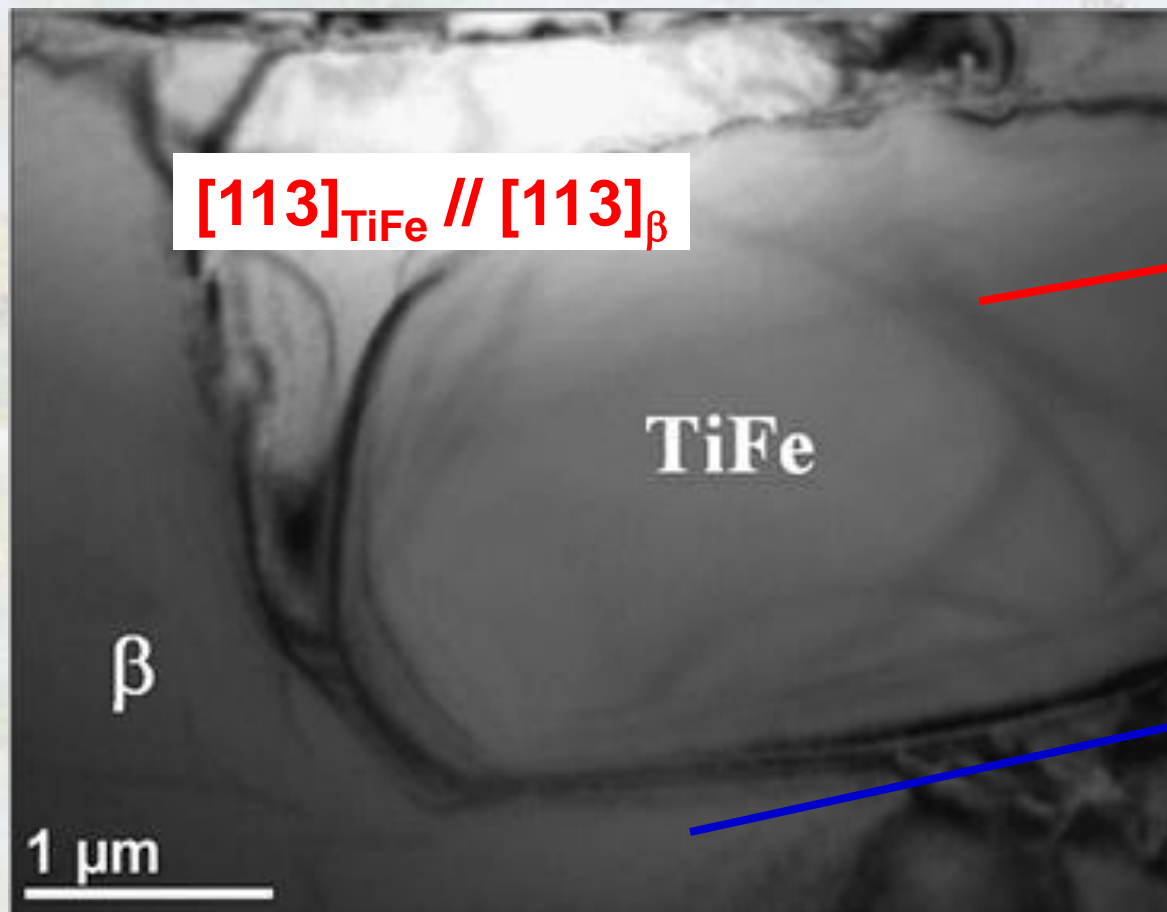
XRD Patterns

- XRD patterns of Ti–Fe eutectic alloys in as-cast and directionally solidified (DS) conditions



TEM Analysis

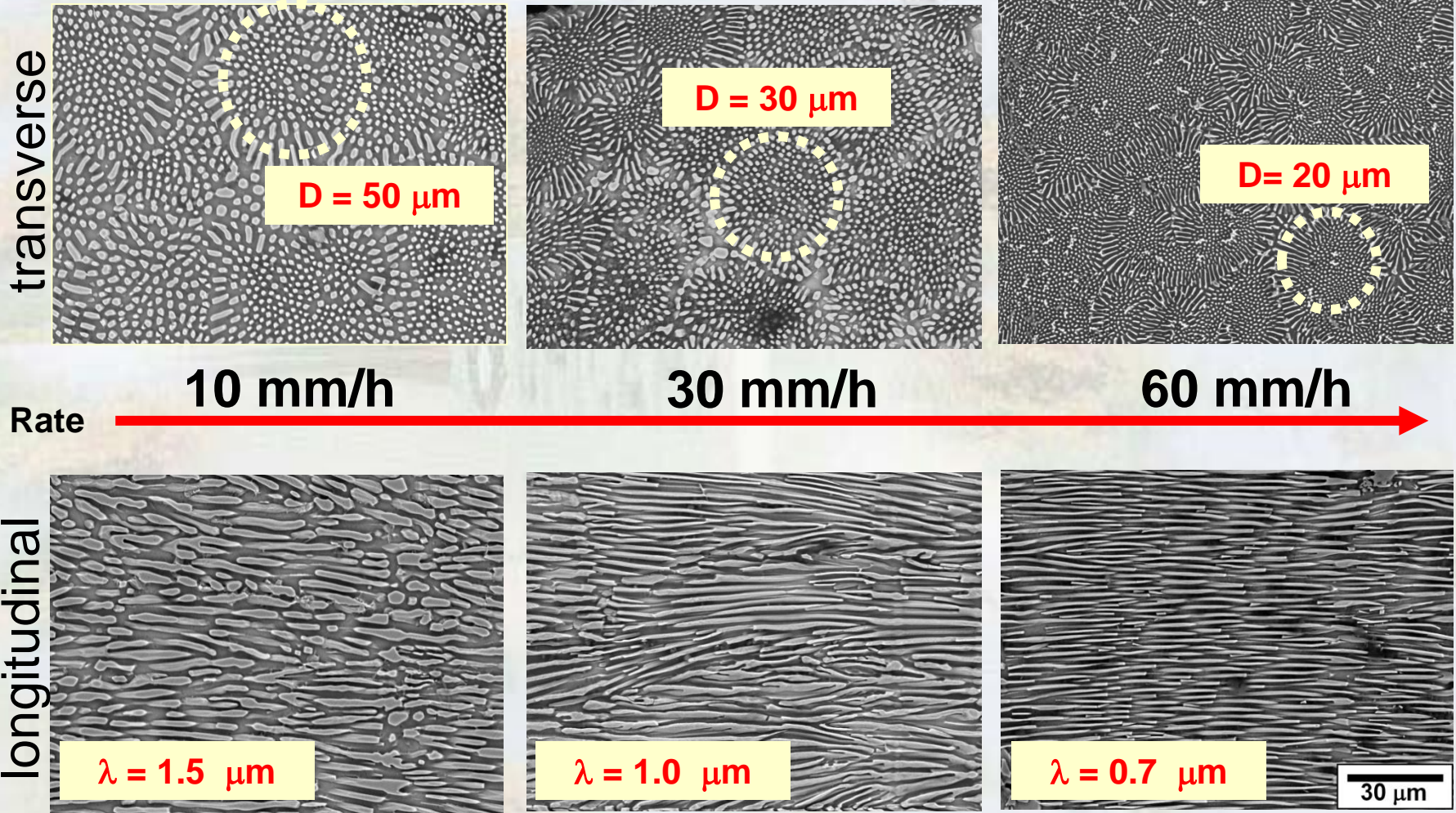
- TEM micrographs in bright field mode and SADP





Solidification Rate

- SEM micrographs showing transverse and longitudinal cross-sections at different rates

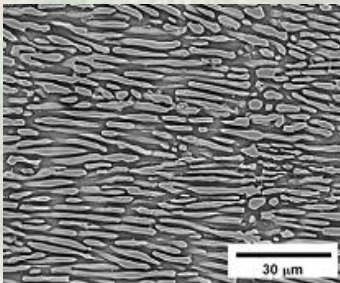




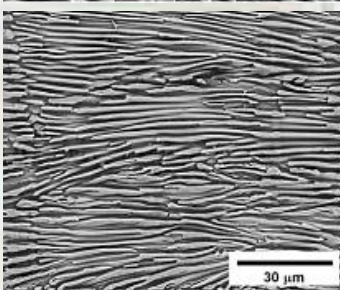
Solidification Rate

- Relationship between the average interspace and the solidification rate of d.s. Ti-Fe eutectic alloy

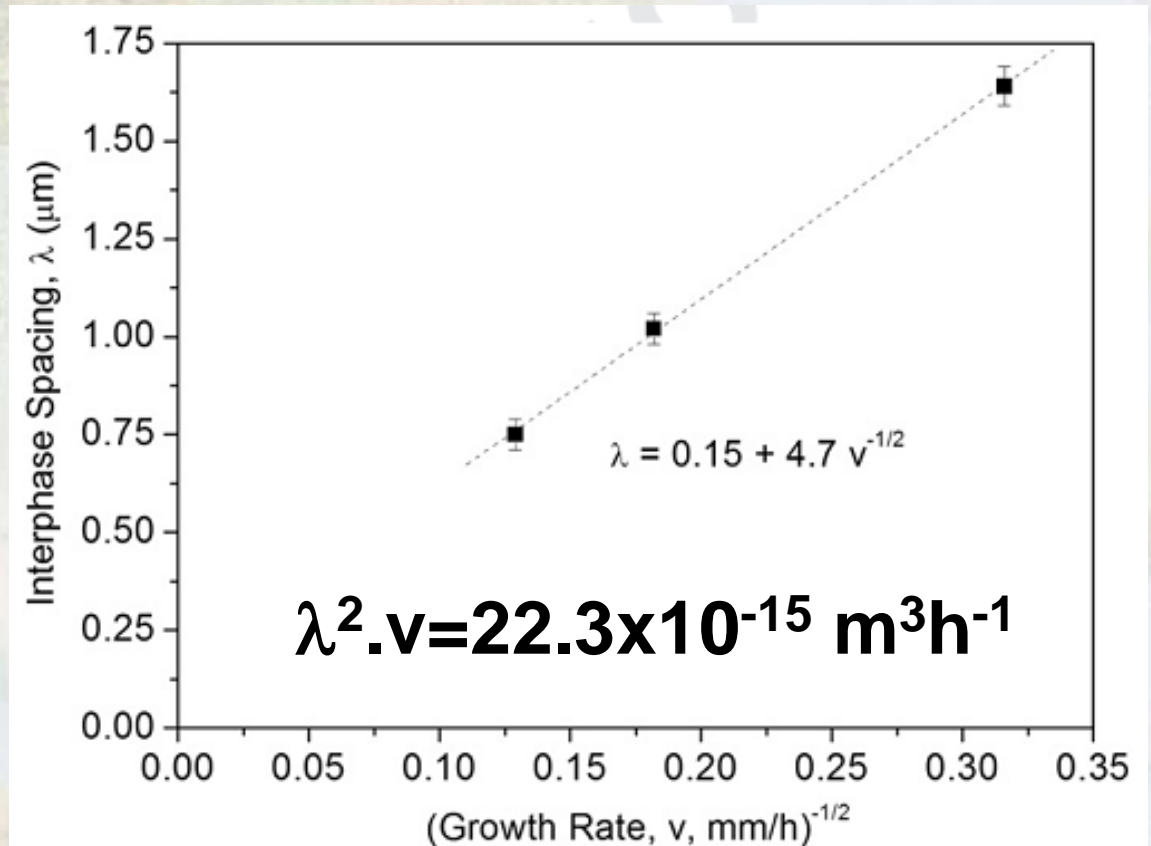
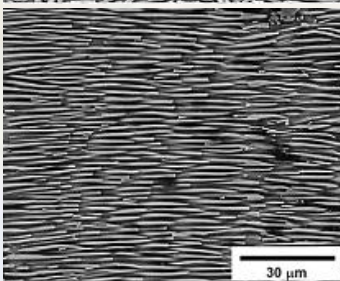
10 mm/h



30 mm/h



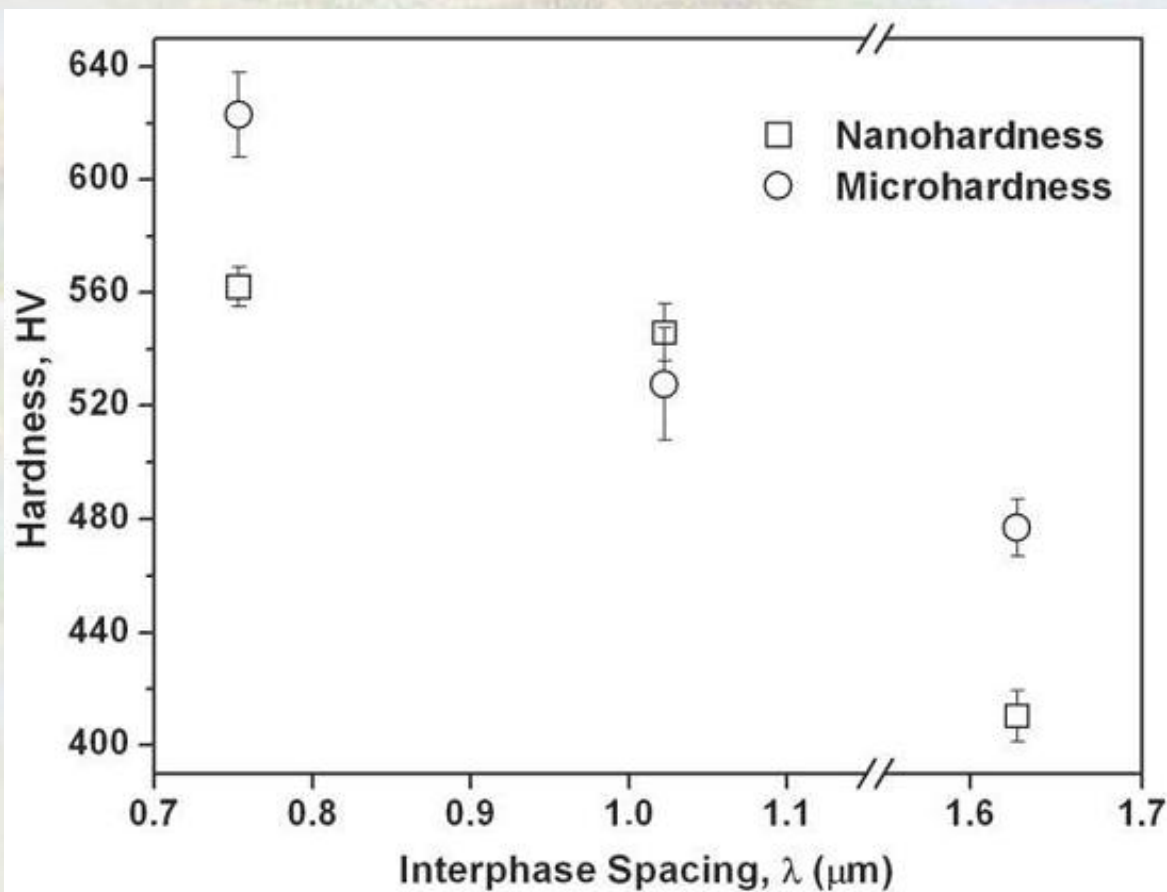
60mm/h





Mechanical Tests

- Evolution of hardness with interphase spacing versus Vickers microhardness (HVmicro) and nanohardness (HVnano).



Nano-indentation:
Three-sided Berkovich diamond indenter and applying a maximum load of 500 Mn:

$$E_E = 110 \text{ to } 177 \text{ GPa}$$

$$E_{\text{TiFe}} = 137 \text{ GPa}$$



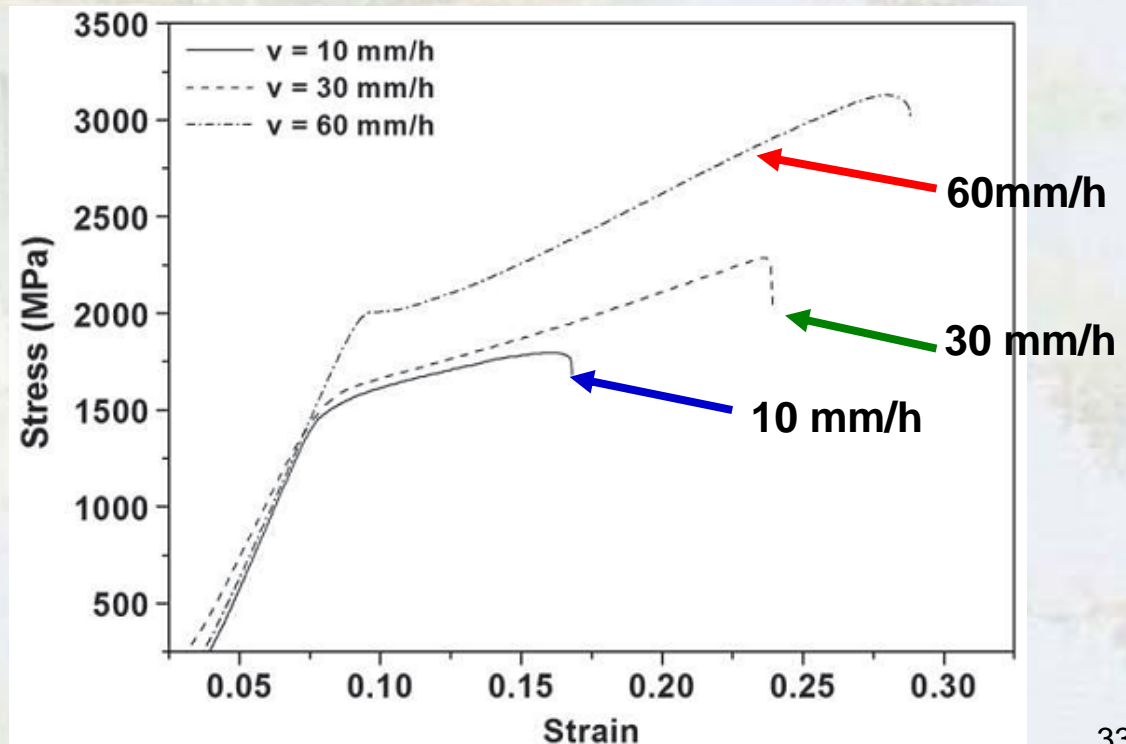
Mechanical Tests

- Compressive mechanical properties

λ (μm)	Ultimate stress σ_{max} (MPa)	Yield strength σ_y (MPa)	Yield strain ϵ_y (%)	Fracture strain ϵ_f (%)
0.7	3000 ± 137	1926 ± 69	11.7 ± 1	25.2 ± 2
1.0	2494 ± 161	1790 ± 124	8.5 ± 1	23.0 ± 1
1.5	1844 ± 101	1644 ± 72	8.1 ± 1	21.6 ± 1

Compression tests:

- Strain rate of $8 \times 10^{-3} \text{s}^{-1}$
- Samples 4 mm high and 2 mm in diameter





Conclusions

New experimental setup was applied to D.S. of Ti-Fe eutectic:

- **No oxygen contamination**
- **No evidence of oxygen rich phase**
- **Well aligned eutectic microstructure**
- **Eutectic transformation at 1080°C**
- **TEM/SADP**
 - **orientation relationship: $(113)_\beta \parallel (113)_{\text{TiFe}}$**
- **$\lambda^2 v = 22.3 \times 10^{-15} \text{ m}^3/\text{h}$**
- **E_E varies from 110 to 177 GPa**
- **σ_{UTS} varies from 1844 to 3000 MPa**
- **Ductility varies from 21.5 to 25.2 %**



Acknowledgments

- **The State of São Paulo Research Foundation**
- **The Brazilian National Council for Scientific and Technological Development**
for financial support



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